

# CUCURBITURIL-FULLERENE COMPLEX

## BACKGROUND OF THE INVENTION

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### Field of the Invention

The present invention relates to a complex composed of cucurbituril and fullerene and a method for manufacturing the complex on a solid-phase.

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### Description of the Related Art

Cucurbituril contains hard cavities, and thus it can be properly used as a host for small organic compounds. In fact, an 1:1 complex or an 1:2 complex of cucurbit[6]uril(or cucurbit[7]uril) and a guest material has been introduced, and concretely a complex of cucurbit[7]uril and 4,4'-bipyridinium-dication, produced in an aqueous solution containing alkaline-metal salt, has been reported(Mock, W. L. and Shih, N. -Y., *J. Org. Chem.* 1986, 51, 4440; Ong, W. et al., *Org. Lett.* 2002 4, 1791; and Kim, H. -J. et al., *PNAS* 2002, 99, 5007). Different from cyclodextrin, a carbonyl group, located at the entrance of a cavity of

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cucurbituril, can be coupled with polarizable molecules and ions by ion-bipolar reaction and/or hydrogen bonding(Mock, W. L., *Top. Curr. Chem.* 1995, 175, 1; and Cintas, P. J., *J. Inclusion Phenom. Mol. Recognit. Chem.* 1994, 17, 205). These complexes are generally produced in a solution, or produced by two-phase reaction between solid-phase cucurbituril and liquid- or gas-phase guest molecules(Mock, W. L. and Shih, N. -Y., *J. Org. Chem.* 1986, 51, 4440; and Buschmann, H. -J. et al., *European Water Pollution Control* 1996, 6, 21).

Cucurbit[6,7]uril is a very desirable composition unit of an acceptor-supramolecule compound composition(Jeon, Y. -M. et al., *J. Am. Chem. Soc.* 1996, 118, 9790; and Zhao, J. et al., *Angew. Chem. Int. Ed.* 2001, 40, 4233), however, its characteristics and applicability have not been studied very much. Up to the present, guest materials that have been used for manufacturing a cucurbituril-containing complex are limited to iodine(Saenger, W., *Angew. Chem.* 1980, 92, 343), dyes for water-purification(Buschmann, H. -J. et al., *European Water Pollution Control*, 1996, 6, 21), and alkyl-and aryl-ammonium ions(Mock, W. L.; Shih, N. -Y.; *J. Org. Chem.* 1986, 51, 4440). And, it has not been reported about a complex of cucurbituril and

nonpolar guest material.

In addition, fullerene has a very high reactivity to a free radical(Krusic, P. J. et al., *Science* 1991, 254 1183; Geckeler, K. E. and Arsalani, N., *Fullerene Sci. Technol.* 1996, 4, 897; and Ford, W. T. et al., *Macromolecules* 1997, 30, 6422) and is able to cut DNA with light being existed, and thus it has very high applicability for bio-medicines. Consequently, it has been broadly researched on a host-guest material chemistry of [60]fullerene. It has been reported that various molecules such as  $\beta$ -and  $\gamma$ -cyclodextrin' (Murthy, C. N. and Geckeler, K. E., *Chem. Commun.* 2001, 1194; Murthy, C. N. and Geckeler, K. E., *Full., Nanotubes, & Carb. Nanostructures* 2002, 10(2), 91; and Anderson, T. et al., *Chem. Soc. Chem. Commun.* 1992, 604), calix[3,5,6,8]arenes(Ikeda, A. et al., *J. Am. Chem. Soc.* 1999, 121, 4296; and EP0686644), and porphyrin-metal macro-ring(Tashiro, K. et al., *J. Am. Chem. Soc.* 1999, 121, 9477) are working as a host for [60]fullerene respectively.

#### SUMMARY OF THE INVENTION

The inventors of the present invention conceived the broad applicability of cucurbituril and fullerene.

So, we tried to produce a complex of cucurbituril and fullerene, a nonpolar material. As a result, we developed a novel complex composition method of manufacturing a complex by solid-phase reaction that is very easy and effective.

It is an object of the present invention to provide a complex of cucurbituril and fullerene.

It is another object of the present invention to provide a method for manufacturing a complex of cucurbituril and fullerene.

To achieve the object mentioned above, the present invention provides a cucurbituril-fullerene complex in which cucurbituril and fullerene are being coupled by molecular interaction.

To achieve the another object mentioned above, the present invention provides a method for manufacturing a complex containing cucurbituril and fullerene that includes crushing a mixture of solid-phase cucurbituril and fullerene.

#### BRIEF DESCRIPTION OF THE DRAWINGS

FIG. 1 is a view illustrating the coupling state of a cucurbit[7]uril-[60]fullerene complex.

FIG. 2 is a graph showing the X-ray diffraction

analysis results of (a)cucurbit[7]uril-[60]fullerene complex, (b)cucurbit[7]uril and (c)[60]fullerene.

FIG. 3 is a graph showing the FT-IR spectrums of (a)cucurbit[7]uril-[60]fullerene complex, (b)cucurbit[7]uril and (c)[60]fullerene.

FIG. 4 is a graph showing the thermo-gravimetric analysis results of cucurbit[7]uril-[60]fullerene complex and cucurbit[7]uril.

#### DETAILED DESCRIPTION OF THE EMBODIMENTS

Hereinafter, the present invention is described in detail.

A complex in accordance with the present invention is a supramolecular complex comprising cucurbituril and fullerene. The cucurbituril used in the present invention is not deformed and comprises cucurbit[6]uril or cucurbit[7]uril. All kinds of fullerenes fit for the cavities of said cucurbituril, such as [60]fullerene, [70]fullerene, etc., can be used in the present invention.

The fullerene used in the present invention is nonpolar material and coupled at the entrance of the cavity of cucurbituril entirely by molecular interaction, not by covalent bonding, to form a stable

complex.

The molar ration of the initial compounds, Cucurbit[7]uril and C60, ranged between 1:2 and 2:1. In all cases the formation of a CUC7:F6=1:2 complex  
5 could be observed.

Fullerene, used in the present invention, has potential characteristics as a bio-medicine such as the functions of eliminating free radicals, cutting DNA, and the like. And cucurbituril is working as  
10 acceptor or absorption material of fullerene. Thus, a complex in accordance with the present invention can be used as a medicine delivery means in the field of pharmaceuticals.

The complex in accordance with the present  
15 invention has a single-phase and can be obtained by solid-phase reaction. In other words, a complex in accordance with the present invention can be obtained by crushing a mixture of solid-phase cucurbituril and fullerene.

In more detail, a complex in accordance with the  
20 present invention can be obtained by mixing solid-phase fullerene and solid-phase cucurbituril with molar ratio from 1:2 to 2:1, preferably with 2:1, and crushing the mixture in a mixing crusher under the  
25 room temperature, preferably in a chrome steel mixing

crusher with chrome steel crushing balls being added, with the rotation speed from 20 rpm for about 1 to 10 hours.

Hereinafter, an embodiment of the present invention is described in detail.

Here, the embodiment is only for an example of the present invention, and the present invention is not limited to the embodiment.

[Embodiment 1] - Manufacturing a complex of cucurbit[7]uril and [60]fullerene.

In a typical experiment, a complex was produced by crushing a mixture of 20.1mg ( $28 \times 10^{-3}$  mmol) of [60]fullerene and 16.3mg ( $14 \times 10^{-3}$  mmol) of cucurbit[7]uril (CB[7]) in a chrome steel mixing crusher using chrome steel crushing balls. The crushing was being carried out with the speed of 20rpm for 1 to 10 hours. After washing-out the produced CB[7]-C<sub>60</sub>fullerene complex with warm water, we added 2M of NaOH to the solution to control its pH to be 12 and added 20ml of toluene thereto to dissolve the remaining CB[7] and non-coupled [60]fullerene. After dissolving excessive initial compounds by agitation the mixture for 30 minutes, we allowed the complex to precipitate. The aqueous phase containing the

insoluble complex was frozen, so that the upper organic phase could be decanted. Next, after leaving the aqueous-phase until it gets back to room temperature, we centrifuged it under 0°C with 5000rpm for 10 minutes, and then poured out the water carefully. After washing the complex with pure water until its pH got to be neutral, we finally evaporated the remained water and vacuum-dried the dark-brown complex to obtain the complex of the present invention(yield rate: 33%).

Referring to the X-ray diffraction analysis results shown in FIG. 2, the produced complex did not show the typical  $2\theta$  values of cucurbit[7]uril and [60]fullerene. Like the cases observed in other complexes containing [60]fullerene, it is shown that the crystal structures of initial compositors, i. e., cucurbit[7]uril and [60]fullerene, are concealed, by forming a complex, in the complex of the present invention.

Referring to the FT-IR spectrums shown in FIG. 3, a characteristic absorption band of cucurbit[7]uril and a typical absorption band of  $C_{60}$ fullerene are shown at  $527\text{cm}^{-1}$ . That is to say, it represents that the complex obtained in the present invention comprises cucurbit[7]uril and  $C_{60}$ fullerene.

Referring to the thermo-gravimetric analysis



results shown in FIG. 3, the total weight loss of the complex of cucurbit[7]uril and [60]fullerene was 40.1% at 410°C, and this represents that the weight ratio of cucurbit[7]uril and [60]fullerene in the complex is 1:2.

As mentioned thereinbefore, a complex in accordance with the present invention is a Supramolecular complex comprising fullerene having potential characteristics as a bio-medicine and cucurbituril working as absorption material or acceptor of fullerene. Since the complex in accordance with the present invention can be easily manufactured and handled, it can be usefully used as a medicine delivery means in the field of pharmaceuticals.